Molecular Physics: An International Journal at the Interface Between Chemistry and Physics

Publication details, including instructions for authors and subscription information:
http://www.tandfonline.com/loi/tmph20

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Published online: 14 Jun 2013.

To cite this article: John L. Bohn & Goulven Quéméner (2013) Dipolar radicals in crossed electric and magnetic fields, Molecular Physics: An International Journal at the Interface Between Chemistry and Physics, 111:12-13, 1931-1938, DOI: 10.1080/00268976.2013.783721

To link to this article: http://dx.doi.org/10.1080/00268976.2013.783721

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INVITED ARTICLE
Dipolar radicals in crossed electric and magnetic fields

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(Received 10 January 2013; final version received 5 March 2013)

Paramagnetic, dipolar Hund’s case-a radicals are considered in the presence of arbitrary, non-collinear combinations of electric and magnetic fields. The field-dependent part of the Hamiltonian is found to exhibit a good quantum number, consisting of the projection of the molecule’s total angular momentum along a space-fixed axis that is determined by both the fields and the electric and magnetic dipole moments of the molecule. This quantity remains good even when the fields are non-collinear. Exploiting this feature identifies a set of quantum numbers for the molecule in crossed fields. We dub this set a ‘Hund’s case-X’ basis.

Keywords: Hund’s cases; molecules in electric and magnetic fields; ultracold molecules

1. Introduction

The concept of Hund’s angular momentum coupling cases is useful for naming and organising the energy levels of molecules. Quite generally, the molecular Hamiltonian can be written as a sum of several pieces, \( H = H_1 + H_2 + H_3 + \ldots \), which may fail to commute among themselves. Hence the eigenstates of \( H \) cannot be labelled by the quantum numbers appropriate to each \( H_i \) simultaneously. However, often one of the terms, let us say \( H_1 \), dominates over the others. It is then worthwhile to express eigenstates of \( H \) in terms of the eigenstates and quantum numbers of \( H_1 \), whereby contributions off-diagonal in these states, arising from \( H_2, H_3, \ldots \) are perturbations. In this way, while the quantum numbers of \( H_1 \) are not strictly ‘good’ quantum numbers, they are ‘good enough’: they serve to classify the states, identify characteristic energy level spacings and provide approximate line strengths for transitions [1].

For a molecule immersed in either an external electric or magnetic field, one such quantum number is the projection \( m \) of the molecule’s total angular momentum onto the field axis. This is in fact a rigorously good quantum number, and one that describes the joint system of molecule-plus-field. However, a molecule that is both dipolar and paramagnetic, such as OH, can respond to both electric and magnetic fields. If these fields are not collinear, then rotational invariance of the Hamiltonian is broken, and neither the \( m \) quantum number referred to the electric field axis, nor the one referred to the magnetic field axis, remains good.

Our main point here is the following. For a molecule, like OH, which is represented by Hund’s case-a in the absence of fields, both the electric \( \vec{d} = d\hat{n} \) and magnetic \( \vec{\mu} = \mu\hat{n} \) moments can be regarded, to a good approximation, as collinear with the molecular axis \( \hat{n} \). In combined electric \( \vec{E} \) and magnetic \( \vec{B} \) fields, the field Hamiltonian consists of Stark and Zeeman terms,

\[
H_{\text{field}} = H_S + H_Z = -\vec{d} \cdot \vec{E} - \vec{\mu} \cdot \vec{B} = -\hat{n} \cdot (d\vec{E} + \mu\vec{B}) \tag{1}
\]

Geometrically, this Hamiltonian describes a generalised ‘moment’ \( \hat{n} \) interacting with a ‘combined field’ that is a weighted linear combination of the electric and magnetic fields. The combined field determines an axis of symmetry with respect to which meaningful \( m \) quantum numbers can again be assigned. These are the ‘good enough’ quantum numbers in this situation, and define therefore a kind of Hund’s case, useful even when other effects such as \( \Lambda \)-doubling are considered.

In the following we elaborate on this idea, showing various examples for the OH molecule and the di-halogen ICl. Understanding the behaviour of OH in crossed electric and magnetic fields has suddenly increased in importance, given recent experiments in which trapped gases of this radical, at mK temperatures, experience widely varying relative magnitudes and orientations of the fields [2,3].

2. Formulation

We begin with a Hund’s case-a molecule in a field, described in a given electronic state by the effective Hamiltonian

\[
H = H_{S0} + H_{\text{rot}} + H_S + H_Z + H_{\Lambda} + \ldots, \tag{2}
\]
which represent, in order, the spin-orbit, rotational, fields and $\Lambda$-doubling contributions. We will assume all other effects are perturbative and can be included as necessary. Hund’s case-a brings with it not only a set of quantum numbers, but also a hierarchy of quantum numbers. The primary ones are those pertinent to $H_{SO}$, namely, the signed projections of electronic orbital ($\lambda$) and spin ($\sigma$) angular momenta, and their sum $\omega = \lambda + \sigma$. We will assume throughout that these quantum numbers are well-defined in the electronic state of interest, e.g. the $2\Pi_{3/2}$ ground state of OH or the $A^1\Pi_1$ state of ICl.

Given the value of $\omega$, a secondary quantum number $j$ describes the rotational eigenstates generated by $H_{\text{rot}}$. The value of $j$ cannot be set independently of $\omega$, but is rather contingent on $\omega$, since it must satisfy $j \geq |\omega|$. Finally, for a given value of $j$, the value of $m$ is contingent on both the value of $j$, via the usual restriction $-j \leq m \leq j$, and on the space-fixed axis used to quantise the angular momentum, which gives it a concrete meaning. As alluded to above, this choice of axis is usefully specified by the direction of a single field. Thus the Hund’s case-a basis set is indexed by a particular collection of meaningful quantum numbers:

$$|\lambda \sigma \omega jm\rangle,$$

where the first ket describes the electronic degrees of freedom in the body frame, and the second ket describes the distribution of the molecular orientation in this frame, via

$$|\omega jm\rangle = \sqrt{\frac{2j+1}{8\pi^2}} D_{j\omega}^m(\alpha \beta \gamma),$$

where $(\alpha \beta \gamma)$ are the Euler angles relating the molecular axis $\vec{n}$ to the laboratory-fixed quantisation axis.

As an aside, we note that this basis can be transformed so as to diagonalise the $\Lambda$-doublet Hamiltonian $H_{\Lambda}$, by constructing parity eigenstates that are linear combinations of the states $|+\omega\rangle$ and $|-\omega\rangle$. However, as we are mostly concerned here with states in large electric fields, it is more appropriate to use the states with signed values of $\omega$. To this end, we will denote the magnitude of $\omega$ as $\tilde{\omega} = |\omega|$ where necessary.

The application of electric and magnetic fields will mix different values of rotational quantum number $j$. Nevertheless, its projection $m$ remains good in the presence of fields. To illustrate our definition of $m$ in the crossed field case, we will explicitly deal at first with the low-field limit, where $dE$ is sufficiently smaller than the rotational constant $B_\omega$, and $j$ is a good quantum number. Note that our primary example, OH, satisfies this criterion for experimentally relevant field ranges [2,3]. In Section 3.3, we will relax this assumption allowing different $j$ values to mix. Nevertheless, the construction of a conserved $m$ quantum number will still hold in that case.

### 2.1. Electric fields

The molecule is assumed to be polar, with electric dipole moment $\vec{d} = d\vec{n}$, where $\vec{n}$ denotes the molecular axis. In an electric field $\vec{E}$, the molecule experiences a Stark energy

$$H_S = -\vec{d} \cdot \vec{E} = -dE \cos(\beta),$$

where $E \cos(\beta)$ is the projection of the field on the molecule’s axis, and $\beta$ is the angle between field and dipole.

This operator has no explicit dependence on electron coordinates, and so the electronic matrix element is unity. Moreover, let us consider a field sufficiently weak that the angular momentum $j$ is nearly conserved. The matrix elements of the Stark Hamiltonian are then

$$\langle \lambda \sigma | \langle \omega jm | H_S | \omega jm \rangle | \lambda \sigma \rangle = -dE \langle \lambda \sigma | \langle \omega jm | \cos(\beta) | \omega jm \rangle \rangle = -dE \langle \omega jm | \cos(\beta) | \omega jm \rangle \rangle$$

The remaining matrix element has a standard form [4]:

$$\langle \omega' jm' | \cos(\beta) | \omega jm \rangle = (-1)^{m'-m} (2j+1) \left( \begin{array}{ccc} j & 1 & j \\ -\omega' & 0 & \omega \end{array} \right) \left( \begin{array}{ccc} j & 1 & j \\ -m' & 0 & m \end{array} \right).$$

This expression is of course diagonal in $m$, for the quantisation axis $\hat{E}$. It is also diagonal in $\omega$, reminding us that the signed value of $\omega$ is a good quantum number in the presence of the field. Indeed, re-writing this matrix element in terms of the Wigner–Eckart theorem,

$$\langle \omega jm | T_0^\dagger(\hat{n}) | \omega jm \rangle = (-1)^{j-m} \sqrt{2j+1} \langle \omega jm | T_1^\dagger(\hat{n}) | \omega jm \rangle \left( \begin{array}{ccc} j & 1 & j \\ -m & 0 & m \end{array} \right).$$

identifies the reduced matrix element as

$$\langle \omega jm | T_1^\dagger(\hat{n}) | \omega jm \rangle = (-1)^{j-m} \sqrt{2j+1} \left( \begin{array}{ccc} j & 1 & j \\ -m & 0 & m \end{array} \right),$$

where $\cos(\beta)$ is expressed explicitly as the zero-th component of a first-rank tensor operator $T_0^\dagger(\hat{n})$. Substituting formulas for the $3j$ symbols, the reduced matrix element becomes [4]

$$\langle \omega jm | T_1^\dagger(\hat{n}) | \omega jm \rangle = \frac{\omega}{\sqrt{j(j+1)}} = \cos(\hat{n} \cdot \hat{j}).$$

In this last line, we take the semi-classical approach, and identify this quantity as the mean angle between the molecular axis and the total angular momentum. The quantum
number \( \omega \) thus identifies the direction of the dipole moment relative to the molecule’s total angular momentum. Doing the same for the electric field factor, we can write Stark matrix elements in the form

\[
E_{\lambda}(\omega, m) = -\left( d \frac{\omega}{\sqrt{j(j+1)}} \right) \left( E \frac{m}{\sqrt{j(j+1)}} \right). \tag{11}
\]

This expression factors into a part that depends on the internal workings of the molecule (including the dipole moment), and a part that depends on its relation with the external field. This Hamiltonian is diagonal in the \( m \) quantum number, provided that \( m \) refers to quantisation along the field axis \( \vec{E} \).

### 2.2. Magnetic fields

Similarly, a case-a molecule with electronic spin will experience a Zeeman shift in a magnetic field, given by

\[
H_{\lambda} = -\mu \cdot \vec{B}, \tag{12}
\]

where the magnetic moment is given (in Hund’s case-a) in the body frame of the molecule as

\[
\mu = -\mu_0(\lambda + 2\sigma), \tag{13}
\]

where \( \mu_0 \) is the Bohr magneton, and \( \lambda \) and \( \sigma \) can in principle point in any direction. However, in a good Hund’s case-a molecule, these vectors have vanishing (or very small) contributions in directions orthogonal to the molecular axis. Therefore, in the \( |\lambda\sigma\rangle \) electronic basis, they are replaced by their quantum numbers, and the magnetic moment is assumed to lie parallel to the molecular axis.

As before, the Hamiltonian then depends on the projection of magnetic field on the molecular axis, \( \vec{B} \cos(\beta) \) where now \( \beta \) is the angle between the molecular axis and the magnetic field. The matrix elements of the Zeeman Hamiltonian are therefore

\[
\langle \lambda\sigma|\omega jm|H_{\lambda}|\omega jm\rangle|\lambda\sigma\rangle = \mu_0 B(\lambda\sigma|\omega jm|\cos(\beta)|\omega jm\rangle = (\lambda + 2\sigma)\mu_0 B|\omega jm|\lambda\sigma\rangle|\omega jm\rangle. \tag{14}
\]

This has exactly the same form as Equation (6), but with a dipole moment that depends on the values of \( \lambda \) and \( \sigma \), which modify the reduced matrix element. This is the sense in which, in case-a, both dipoles are proportional to \( \hat{n} \), with the proportionality constant being simply a reduced matrix element that expresses details (e.g. electric or magnetic) inside the molecule. The act of orienting \( \hat{n} \) with respect to the external field is one of pure geometry, and described by the matrix element of \( \cos(\beta) \). Proceeding as above, we can factor the energy into internal and external pieces:

\[
E_{\lambda}(\omega, m) = \left( \mu_0(\lambda + 2\sigma)\frac{\omega}{\sqrt{j(j+1)}} \right) \left( B \frac{m}{\sqrt{j(j+1)}} \right). \tag{15}
\]

This Hamiltonian is diagonal in the \( m \) quantum number, provided that \( m \) refers to quantisation along the field axis \( \vec{B} \).

### 2.3. Crossed fields

Suppose now that the molecule experiences both electric and magnetic fields, which may point in different directions. The field part of the Hamiltonian reads

\[
H_{\lambda} + H_{\sigma} = -\vec{d} \cdot \vec{E} - \mu \cdot \vec{B}. \tag{16}
\]

There is now no obvious quantisation axis – or is there? In a particular electronic state \( |\lambda\sigma\rangle \), the field Hamiltonian in the molecular orientation degree of freedom is

\[
\langle \lambda\sigma|H_{\lambda} + H_{\sigma}|\lambda\sigma\rangle = -d(\hat{n} \cdot \vec{E}) + (\lambda + 2\sigma)\mu_0(\hat{n} \cdot \vec{B}). \tag{17}
\]

This expression has exactly the form of the dot product between an effective ‘moment’ \( \hat{n} \) – identifying the orientation of the molecule – and an effective field that combines the electric and magnetic fields. There are actually two such combined fields, according to the sign of the zeroth-order \( g \)-factor, \( g = \lambda + 2\sigma \). Thus there are two distinct field Hamiltonians

\[
H_{\kappa} = -\hat{n} \cdot \vec{C}_\kappa, \tag{19}
\]

where

\[
\vec{C}_\kappa = d\vec{E} + \kappa|g|\mu_0\vec{B} \tag{20}
\]

with \( \kappa = \pm 1 \). This sign convention implies that \( \kappa = +1 \) stands for an ‘energetically stretched’ state. That is, for parallel \( \vec{E} \) and \( \vec{B} \) fields, both fields shift the energy in the same direction. Thus the electric and magnetic moments align in the same direction in the body frame of the molecule, implying in turn that \( \lambda + 2\sigma \) is negative. This means that in general \( \kappa \) has the opposite sign to \( \omega \). The geometric construction of the combined fields is illustrated in Figure 1.
An alternative sign convention would give κ the same sign as ω = λ + σ itself. To make this identification in all cases would be, however, potentially ambiguous: in a $^3Π_0$ state, for instance, $ω = 0$ and cannot serve as a signed quantum number (even though $g$ is non-vanishing). Likewise, the signed value of $g = λ + 2σ$ is not necessarily helpful, as it is nominally zero for states such as $^2Π_{1/2}$. The actual $g$-factor of course can be non-zero, but its sign can be difficult to determine without detailed consideration of the molecule. For these reasons, κ emerges as a new quantum number, with obvious ties to $ω$.

For a particular internal state identified by λ and σ (therefore, κ is determined), $\hat{C}_κ$ serves as a quantisation axis. By analogy with the above, the Hamiltonian becomes

$$\langle \lambda\sigma |(ωjm_κ)(H_\text{S} + H_\text{Z})(ωjm_κ)|\lambda\sigma \rangle = -C_κ(ωjm_κ)\cos(β)(ωjm_κ),$$

(21)

where $\hat{C}_κ$ is the angle between $\hat{n}$ and $\hat{C}_κ$, and the subscript on $m_κ$ emphasises the angular momentum projection onto the $\hat{C}_κ$ axis. From here, the problem is mathematically equivalent to the results above. In particular, the exact energy spectrum of the field Hamiltonian is given by

$$E_{\text{S+Z}}(ω, m) = \left(\frac{ω}{\sqrt{j(j+1)}}\right)\left(C_κ \frac{m_κ}{\sqrt{j(j+1)}}\right),$$

(22)

where $C_κ$ is the magnitude of the combined field,

$$C_κ = \sqrt{(dE)^2 + (gμ_0B)^2 + 2κd|gμ_0B|cos(θ_{EB})},$$

(23)

and $θ_{EB}$ is the angle between the electric and magnetic fields. Significantly, Equation (22) is diagonal in the $m_κ$ quantum number, provided that $m_κ$ refers to quantisation along the field axis $\hat{C}_κ$ for a particular value of $κ$. In this sense $κ$ denotes another quantum number of the combined field-molecule system, on which others are contingent.

The crossed-field case can therefore be solved exactly, and quantum numbers can be assigned to the different energy levels, for electric and magnetic fields of arbitrary strength and relative orientation. The way to make this possible is to accept that the quantum numbers are now conditional, that is, $m_κ$ cannot be assigned unambiguously until the internal state $ω$ (and hence $κ$) are specified. Because these states identify good quantum numbers in the crossed field case (suggested by the letter ‘X’), we refer to this basis as the Hund’s case-X coupling scheme, with basis sets denoted $|ωkjm_κ\rangle$.

### 2.4. Lambda doubling

We have deliberately focused on the situation where the electric field interaction, $dE$, is larger than the $λ$-doublet splitting $Δ$ in the molecule. This has ensured that the signed values of $ω$ are good quantum numbers, rather than the parity quantum number $σ_p$ in the parity states $|ω0⟩ |e_p⟩ - |−ω⟩ |e_p⟩ /\sqrt{2}$. To complete the picture, we must construct matrix elements of $λ$-doubling in our basis.

Starting with the case of zero magnetic field, the Hamiltonian consists of Stark and $λ$-doubling terms

$$H = H_\text{S} + H_λ.$$  

(24)

In the signed basis $|±ωjm⟩$, the Hamiltonian matrix reads

$$H = \left(-dE \frac{ωm_λ}{2(j+1)} + dE \frac{ωm_λ}{2(j+1)} \right),$$

(25)

which gives the familiar eigenvalues

$$\frac{ω}{2\bar{ω}}\sqrt{\left(\frac{dE}{j(j+1)}\right)^2 + \left(\frac{Δ}{2}\right)^2},$$

(26)

where $Δ$ is the zero-field $λ$-doublet splitting. More concisely, non-zero matrix elements of the $λ$-doubling Hamiltonian are given by

$$\langle -λ - σ |(−ωjm)|H_λ|ωjm⟩|λσ⟩ = \frac{Δ}{2}δ_{mm′},$$

(27)

provided $m′$ and $m$ are referred to the same quantisation axis. This form of the Hamiltonian makes evident that the $λ$-doubling connects states of $+\bar{ω}$ to states of $−\bar{ω}$, that is, in the case-X picture it mixes the states $\pm κ$ that refer to different axes $\hat{C}_κ$.

To compute matrix elements of $H_λ$ in the case-X basis, we therefore have to transform between these two axes. For concreteness, denote by $m_±$ quantum numbers referred...
to the $C_\perp$ axes, and let $\Omega_C = (0, \theta_C, 0)$ be the set of Euler angles defining the rotation between these axes, with $\cos(\theta_C) = \hat{C}_+ \cdot \hat{C}_\perp$. Without loss of generality, the plane of the two axes defines the laboratory $x$-$z$ plane, whereby the other two Euler angles can be set to zero. The rotation matrix between the two axes is then denoted $D$, with matrix elements $D_{m_-,m_+}(0, \theta_C, 0) = d_{m_-,m_+}(\theta_C)$ in terms of the Wigner $D$ matrices [4].

Writing the Hamiltonian in block-diagonal form, with the blocks denoting $\kappa = +1$ and $\kappa = -1$ states, the transformation reads

$$
\begin{pmatrix}
1 & 0 \\
0 & D^\dagger
\end{pmatrix}
\begin{pmatrix}
0 & H_A \\
H_A & 0
\end{pmatrix}
\begin{pmatrix}
1 & 0 \\
0 & D
\end{pmatrix} = 
\begin{pmatrix}
0 & H_A D \\
D^\dagger H_A & 0
\end{pmatrix},
$$

whereby the matrix elements of $\Lambda$-doubling in the case-$X$ basis read

$$
\langle \kappa \sigma | -\omega k m_\kappa | H_A | \omega - \kappa m_{-\kappa} | \lambda \sigma \rangle = \frac{\Delta}{2} d_{m_-,m_+}^j (\theta_C).
$$

Matrix elements between states with the same value of $\omega$ (hence connecting states from the same set $m_-$ or $m_+$) vanish.

### 3. Examples and applications

Armed with these analytic results, certain aspects of molecules in the combined fields can be elucidated. In Sections 3.1 and 3.2, we consider explicitly the $^2\Pi_{3/2}$ ground state of the OH radical, admitting only field values sufficiently small that $dE \ll B$, so that the total angular momentum $j = 3/2$ remains a good quantum number. In Section 3.3, we relax this restriction to treat the $^4\Pi_1$ state of the ICl molecule, in which many $j$'s are mixed in the field range considered.

#### 3.1. Magnetic trapping of polar radicals

Cold paramagnetic radicals are amenable to magnetic trapping in mangeticstatic traps, just as, say, alkali atoms are. In principle, this would leave the electric field as an independently variable tool to manipulate and study the electric field response of these dipolar species. However, for case-a molecules like OH, electric and magnetic field effects are confounded. Within the case-$X$ formalism, we can identify the states involved for a particular field configuration.

For example, for colinear fields, $\theta_{EB} = 0$, the energies of the states read

$$
-\frac{m \omega}{j(j+1)} (dE + \kappa |g| \mu_0 B).
$$

This formula emphasises the fact that, for $\kappa > 0$, the magnetic dipole moment points anti-parallel to the electric dipole moment. Thus those states that rise in an electric field rise further in a magnetic field, and those that decrease in an electric field decrease further in a magnetic field, an effect which has long been known [5]. This is shown in the plot of energies versus $B$ in Figure 2(a), where $B$ is parallel to an electric field of magnitude $E = 5 \text{kV/cm}$, for the ground $^2\Pi_{3/2}(j = 3/2)$ state of OH. The solid lines denote the $\kappa = +1$ states. By contrast, for the $\kappa = -1$ states (dashed lines), the magnetic and electric fields pull in opposite directions; higher-energy states go lower, and vice versa, leading to a crossing at $\sim 1500 \text{ Gauss}$. To finish off the picture of parallel fields, Figure 2(b) shows the same energy levels but including the effect of $\Lambda$-doubling (computed in the case-$X$ basis as described above), illustrating that, while it breaks the degeneracy, it is indeed a perturbation.

Figure 2(c) shows the energies of the same OH molecule in the same 5 kV/cm electric field, but with a magnetic field tilted at an angle $\theta_{EB} = \pi/4$ relative to it. Again the $\kappa = +1$ states are denoted by solid lines, while the $\kappa = -1$ states are denoted by dashed lines. In this case the combined field $\vec{C}_\perp$ lies somewhere in the acute angle between the directions of $\vec{E}$ and $\vec{B}$ (Figure 1). Thus, while the fields do not pull the molecular axis in quite the same direction, they almost do so. The effect is that the $\kappa = +1$ states fan out in energy, just as in Figure 2(a).

More interesting are the $\kappa = -1$ states (dashed lines) in Figure 2(c). The degeneracy that was apparent for parallel fields in Figure 2(a) is now gone, replaced by avoided crossings. These crossings arise from the magnetic field breaking rotational symmetry about $\vec{E}$ and therefore mixing states of different $m$ referred to $\vec{E}$. Ordinarily, one would obtain them by a numerical diagonalisation of the Hamiltonian (1). In the case-$X$ picture, each of these states is determined exactly by a unique value of the quantum number $m_-$ for any value of $\vec{B}$. However, the quantisation axis is of course different for each $\vec{B}$. Therefore, the states do change throughout the crossing, and there remain non-adiabatic couplings between the different states.

As before, the effect of including the $\Lambda$-doubling is to perturb these energies somewhat (Figure 2(d)). Doing so of course introduces further crossings between the case-$X$ states. Still, away from the main region of crossings, the case-$X$ quantum numbers ($\kappa$, $m_\kappa$) remain useful for classifying states. Approximate energy eigenvalues for this situation, for spin-1/2 molecules, were extracted from a semi-classical model in Ref. [6]. Interestingly, the full $8 \times 8$ Hamiltonian matrix for this case can be diagonalised analytically [7].

Finally, consider the case where the electric and magnetic fields are perpendicular [Figure 2(e)]. Now the two combined fields $\vec{C}_\pm$ point in opposite directions, but lie along the same line. As a consequence, they define the same quantisation axis. From Equations (22) and (23), it can be seen that, in the absence of $\Lambda$-doubling, each state ($\kappa = +1, m_+ $) is exactly degenerate with the state ($\kappa = -1,$
Figure 2. Zeeman effect for OH molecules in their $^2 \Pi_{1/2}, j = 3/2$ ground state, subject also to an electric field of magnitude $E = 5 \text{kV/cm}$ that makes an angle $\theta_{EB}$ with respect to the magnetic field. Shown is the approximation without including $\Lambda$-doubling (left column, black), and including it (right column, red). In each panel on the left, states with $\kappa = +1$ are drawn using solid black lines, while those with $\kappa = -1$ are drawn using dashed black lines.

$m_\pm = -m_\mp$). Including the $\Lambda$-doubling therefore mixes degenerate states of opposite parity at all values of $B$, hence has a comparatively large influence on the spectra even at large fields [Figure 2(f)].

Incorporating an electric field into the OH magnetic trap also has implications for Majorana transitions in the trap. Consider the magnetic field configuration of a quadrupole trap,

$$\mathbf{B}(\mathbf{r}) = \delta B(x\hat{x} + y\hat{y} - 2z\hat{z}),$$  \hspace{1cm} (31)

where $\delta B$ represents the field gradient. A magnetic moment $\mathbf{\mu}$ that adiabatically tracks this field, and is everywhere parallel to it, experiences a trapping potential

$$U_{\text{trap}}(\mathbf{r}) = -\mathbf{\mu} \cdot \mathbf{\tilde{B}}(\mathbf{r}) = \mu \delta B(x^2 + y^2 + 4z^2)^{1/2}. \hspace{1cm} (32)$$

The problem, of course, is that the magnetic moment cannot track the field adiabatically at $\mathbf{r} = 0$, where the field vanishes. This non-adiabaticity leads to the Majorana losses.

In combined fields, however, the situation is different. Suppose, for example, that the electric field $\mathbf{E} = (E_x, E_y, E_z)$ is constant in the vicinity of $\mathbf{r} = 0$. Then the combined
fields

\[ \tilde{C}_\pm = (dE_x \pm |g|\mu_0 \delta B x) \hat{x} + (dE_y \pm |g|\mu_0 \delta B y) \hat{y} + (dE_z \mp 2|g|\mu_0 \delta B z) \hat{z} \]  

(33)

do not vanish at \( r = 0 \). Now, assuming that the molecular orientation \( \hat{n} \) can adiabatically follow the fields, we have confining potentials

\[ U_{\text{trap, } \pm}(r) = \left[ (dE_x \pm |g|\mu_0 \delta B x)^2 + (dE_y \pm |g|\mu_0 \delta B y)^2 + (dE_z \mp 2|g|\mu_0 \delta B z)^2 \right]^{1/2}. \]  

(34)

This rounding out of the trap minimum may be expected to reduce the rate of Majorana losses.

### 3.2. Electric dipole moments

In some applications, notably cold collisions, the electric dipole moment and its orientation play a decisive role. In a case-X state, the dipole moment \( \vec{d} = \vec{d}_0 \) precesses about the appropriate quantisation axis \( \hat{C}_+ \) or \( \hat{C}_- \) just as it would about the electric field axis in the absence of a magnetic field. The semi-classical direction of the mean dipole \( \langle \vec{d} \rangle \) is therefore unambiguously defined along one of these axes. In applications, however, it may also be useful to relate this direction to the direction of the electric field, to anticipate the role of a magnetic field in re-orienting \( \langle \vec{d} \rangle \) in the lab frame.

To this end, we define the tilt angle, \( \theta_{\text{tilt}} \), between \( \langle \vec{d} \rangle \) and \( \vec{E} \), given by

\[ \cos(\theta_{\text{tilt}}) = \frac{\langle \vec{d} \cdot \vec{E} \rangle}{|\vec{d}| |\vec{E}|}, \]  

(35)

\[ \frac{dE + \kappa |g|\mu_0 B \cos(\theta_{EB})}{\sqrt{(dE)^2 + (g\mu_0 B)^2 + \kappa d |g|\mu_0 E B \cos(\theta_{EB})}}. \]  

(36)

This tilt angle is shown in Figure 3 versus electric field, for OH in a \( B = 1000 \) Gauss magnetic field (and neglecting \( \Lambda \)-doubling). The angle between the fields is arbitrarily set at \( \theta_{EB} = \pi/3 \). At large electric field, \( \theta_{\text{tilt}} \) goes to zero; a strong electric field of course polarises the dipole along itself, regardless of the magnitude and orientation of the magnetic field. For smaller electric fields, the magnetic field makes a significant difference in the dipole’s direction. In the limit of zero electric field, it is rather the magnetic field that sets the direction of the electric dipole (neglecting the \( \Lambda \)-doubling). In this limit the angle between \( \langle \vec{d} \rangle \) and \( \vec{E} \) approaches \( \cos(\theta_{\text{tilt}}) = \kappa \cos(\theta_{EB}) \). That is, in this limit \( \theta_{\text{tilt}} = \theta_{EB} \) when \( \kappa = +1 \), and \( \theta_{\text{tilt}} = \pi - \theta_{EB} \) when \( \kappa = -1 \).

### 3.3. Higher fields and pendular states

At electric fields sufficiently high that the Stark energy \( dE \) becomes comparable to, or larger than, the rotational constant \( B_c \), the energy level spectrum qualitatively changes. In the extreme limit of \( dE/Bc \gg 1 \), the molecule is better described as a two-dimensional harmonic oscillator, which description serves as a starting point for perturbatively evaluating energies at finite values of \( dE/B_c \gg 1 \) [8] (In other words, the oscillator quantum numbers describe the Hund’s case appropriate in the high-field limit). These hindered rotor states, dubbed ‘pendular states’ [9,10], have been explored experimentally for both the electric field and magnetic field [11] versions, as well as in combined fields that are either parallel or anti-parallel [12].

Here we merely point out that the case-X classification scheme serves to identify energy levels even in the event that the fields are non-parallel. To this end, the combined field Hamiltonian takes its full, \( j \)-mixing form

\[ \langle \omega k j m_e | H_S + H_Z | \omega k j m_e \rangle = -C_\phi \langle \omega k j | \mu_0 B \cos(\beta) | \omega k j m_e \rangle \]  

\[ = -C_\phi (-1)^m_e \sqrt{2j^* + 1}(2j + 1) \times \left( \begin{array}{c} j^* + 1 \ j \\ -\omega & 0 & \omega \end{array} \right) \left( \begin{array}{c} j \ j^* + 1 \ m_e \\ -m_e & 0 & m_e \end{array} \right). \]  

(37)

Significantly, the definition of the combined fields \( \tilde{C}_x \), and the subsequent conservation of \( m_e \)’s along these axes, is independent of the fact that \( j \) is not conserved in a field. The quantum numbers \( \kappa \) remain as good as before. To the field interaction, we add the rotational Hamiltonian

\[ H_{\text{rot}} = (j^2 - \omega^2)B_c, \]  

(38)
and diagonalise in a suitable basis of $j$ to determine the energy levels.

This procedure is carried out, with results shown in Figure 4 for the $A^3\Pi_1$ state of the ICl molecule [12]. This figure displays the low-lying energies, versus electric field, over a range that shows the transformation between rotor and pendular states, for the low-lying states. The light, blue line in Figure 4(a) is the result in zero magnetic field. In the presence of a magnetic field $B = 3000$ Gauss, tilted at an angle $\pi/4$ with respect to the electric field, degeneracies are broken, leading to independent spectra in Figure 4(b) for $\kappa = +1$ (solid black) and $\kappa = -1$ (dashed black) states.

4. Conclusion

In the presence of crossed electric and magnetic fields, neither field alone serves as a suitable quantisation axis for eigenstates of a case-a molecule. Interestingly, quantisation axes can nevertheless be found, and good quantum numbers $m_\kappa$ defined for the crossed-field situation. The cost of being able to do so is that two quantisation axes must be identified, which naturally divides the eigenstates into two qualitatively different varieties, according to whether the electric and magnetic dipole moments are parallel or anti-parallel. These axes in general also depend upon the electronic state of the molecule through the quantum numbers $\lambda$ and $\sigma$.

Finally, we remark that the combined fields are not necessary for molecules described by Hund's case-b. For these molecules, the electronic spin is sufficiently decoupled from the molecular axis $\hat{n}$ that the usual laboratory-frame quantum numbers can be used. Specifically, the states $|m_s\rangle$ of spin and $|m_m\rangle$ of rotation will diagonalise the Hamiltonian $H_S + H_Z$, provided that $m_s$ is quantised along the magnetic field axis, and $m_m$ along the electric field axis.

Acknowledgements

This study was supported by the Air Force Office of Scientific Research under the Multidisciplinary University Research Initiative Grant No. FA9550-09-1-0588.

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